

# Quantum Physics, Preparation for Tuesday, Mar. 19

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## Study Q10.4 and Q11.1 to Q11.5

You're finishing Q10 and doing Q11, except that I'm recommending you hold off on Q11.6 just because semiconductors is a substantial subject all of its own. Moore is continuing to give you a lot of the important facts about various solutions of Schrödinger's equation without yet having given you Schrödinger's equation. Of course, I have given it to you and the write-up of my presentation is on our course website. Because this class's material is mostly phenomenological facts, not theoretical derivations, you will probably find the problems pretty easy.

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## Presentation

Ren and Rebecca: Using the Pauli Exclusion Principle to discover how much energy it requires to stuff  $2N$  electrons in a 1-D or 3-D box, ignoring Coulomb repulsion

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## For Problem Set 12

### An application of Heisenberg's Uncertainty Principle

1. Q10B.3, p. 163, Moore just wants you to take  $\Delta x$  as 8 fm and determine  $\Delta p$  by solving for it in the Heisenberg Uncertainty Principle. Very hand-waving, and very easy, but still useful.

### An elementary application of Bohr's model of the atom

2. Q10B.6, p. 163, you can use the result quoted in Q10B.4

### A creative application of Bohr's model of the atom — to the harmonic oscillator

3. Q10M.5, p. 163

### A kind of muonium

4. Q10M.6, p. 164, actually the term "muonium" is reserved for a  $\mu^+ - e^-$  atom, but this gets at the idea

### The Pauli Exclusion Principle

5. Q11M.9, p. 180, a proton and a neutron can be in exactly the same state (because they are *intrinsically* different, so it doesn't violate the Pauli Exclusion Principle)